PEROVSKITES. FROM PIGMENTS TO MULTIFERROICS, TO THE DISCOVERY OF THE LOCKED-TILT PEROVSKITES

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Superconductors, non-toxic inorganic pigments, sensitizers in dye-sensitized solar cells (DSSCs), semiconductors, and multiferroics are just a few examples of the broad range of uses for which ceramics with perovskite structure are employed. In its aristotype form the ABO_3 perovskite structure is cubic (s.g. Pm-3m) with A cations located at the center of dodecahedral sites defined by a three-dimensional array of corner-sharing BO_6 octahedra. This polyhedral arrangement makes the perovskite structure extremely flexible. Indeed, due to changes in the chemical nature of A and B cations as well as on temperature and pressure conditions, the ideal cubic symmetry is often lowered, and the resulting structure is described by a combination of tilts and distortion of the BO_6 octahedra (Mitchell, 2002).

In this contribution, a series of case studies on the interplay among structural characterizations, coloration mechanisms, and technological prospects of ceramics with perovskite structure are outlined. Starting from the Crdoped (Y,REE)AlO₃ (YAP) case, *i.e.* the only system manufactured at the industrial scale and currently used as ceramic red pigment (Ardit et al., 2016), the first case study (i.e., a X-ray powder diffraction and electronic absorption spectroscopy combined investigation) is focused on the effects due to the Al \leftrightarrow Cr substitution at the B octahedral site of the YAP structure, which cause a degree of structural relaxation around Cr^{3+} that, whether compared with other Cr-bearing structures, is relatively low (Cruciani et al., 2009). The concept of crystal field stabilization energy CFSE, closely related with that of structural relaxation (Burns, 1993), will lead to the second case study where *in situ* high-pressure synchrotron XRD measurements coupled with *ab initio* simulations of the electronic population of the orthorhombic NdCrO₃ perovskite are compared with the compressional feature of NdGaO₃ (high-tech ceramics applied as interconnector of solid oxide fuel cells (SOFC), substrate for high-*Tc* superconductors (HTSC), colossal magnetoresistive (CMR) film epitaxy, and so on). The different electronic configuration of octahedrally coordinated Cr^{3+} and Ga^{3+} ions that leads to a redistribution of electrons at the 3d orbitals for Cr^{3+} allows the CFSE at octahedral sites to act as vehicle of octahedral softening in NdCrO₃ or it turns octahedra into rigid units when CFSE is null as in NdGaO₃. The third (and last) case study deals with the first finding of a perovskite characterized by absence of changes in the octahedral tilting and a volume reduction with pressure exclusively controlled by an isotropic polyhedral compression (Ardit et al., 2017). A synchrotron structural investigation at high-pressure sets YAl_{0.25}Cr_{0.75}O₃ as the prototype of the so-called *locked-tilt* perovskites. Besides to reveal a new P-T thermodynamic scenario in the world of perovskite compounds, such a freezing of the octahedral rotations offers a new (and antithetic, keeping in mind that oxygen rotations are strongly coupled to the functional properties of perovskite compounds) perspective on the research of new functional materials (e.g., ferroelectrics and multiferroics). For example, the mutual interaction between layers of octahedrally tilted perovskites subjected to an external perturbation can be easily tuned whether one of the layers is a *locked-tilt* perovskite.

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